

PARENTERALS

STORAGE CONDITIONS		IP TERMINOLOGY FOR SOLUBILITY	
STORAGE CONDITION	MEANING	DESCRIPTIVE TERM	APPROXIMATE VOLUME OF SOLVENT IN ml/gm OF SOLUTE.
• Cold storage or under refrigeration	Usually 2 to 8 degree C.	• Very soluble	Less than 1 part
• Cool storage	Bet ⁿ . 8 to 25°C.	• Freely soluble	1 to 10 part.
• Room temperature	20 to 25°C.	• Soluble	10 to 30 parts.
• Warm	Bet ⁿ . 30 to 40°C	• Sparingly	30 to 100 parts.
• Freezer	store bet ⁿ . -5 to -20°C	• Slightly	100 to 1000 parts.
		• Very slightly soluble	1000 to 10,000 parts.
		• Practical in-soluble	More than 10,000

WATER :- Commonly used vehicle for drug solⁿ.

TEST FOR QUALITY OF WATER

- A gravimetric method.
- Electrolytic measurement of conductivity

TYPE	METHOD OF PREPARATION	COMMENT
• WFI	- Distillation or reverse osmosis	Not sterile, used within 24 hrs./stored below 2°C to 30°C
• Purified Water	- Distillation & ion exchange	Pharmaceutical solvent Single dose container. Multiple & single dose.
• Sterile WFI	- Distillation or reverse osmosis	
• Bacteriostatic WFI	- Distillation or reverse osmosis	

NOTE :- Total solid content in WFI is 10 ppm / sterile WFI is 20 to 40 ppm
Conductivity of WFI, should not more than 1 microhm^{-1} .

ROUTES OF ADMINISTRATION		
ROUTE	SITE & QUANTITY	EXAMPLE
• SC	Under skin layer	Insulin.
• ID/IC	into dermis layer vol. 0.1 to 0.2 ml	Tuberculin skin testing etc.
• IV	into vein, volume upto 500 ml.	Only aqueous preparation.
• Intrathecal or Intra-spinal	into CSF or spinal cord vol. less than 20 ml.	spinal anaesthetic & antibiotics.
• Intra-arterial	into artery	Antineoplastic drugs
• Intra-articular	directly into joints	Antibiotics. Morphine, steroids.

TYPES OF PARENTERAL PRODUCTS	
TYPE	VOLUME
• Small vol. Parenteral (SUV)	Vol. < 100 ml or equal to 100 ml
• Large vol. Parenteral (LUV)	Vol. > 100 ml.
• Single dose container	size is limited to 1000 ml
• Multiple dose container	size is limited to 30 ml

PARENTERALS

FORMULATION OF PARENTERALS

VEHICLE	EXAMPLE
Aqueous Vehicle (Water)	NaCl injection, Ringer solution, dextrose solution, Lactated - ringer solution.
Water miscible	Ethyl alcohol, Propylene glycol, PEG 300, 400, 600, Glycerine.
Non-aqueous	Fixed oil - ethyl oleate, isopropyl myristate etc.

* ADDITIVES :- Anti-micro bacterial preservatives

Ex :- Phenyl mercuric nitrate, phenyl mercuric acetate, methyl paraben, propyl paraben, Benzyl alcohol, phenol, chlorobutanol

ANTIOXIDANTS	EXAMPLES
1. Reducing agents (by oxidizing itself).	Ascorbic acid, Sodium bisulphite, Sodium metabisulfite, Thiourea.
2. Blocking agents. (block oxidative chain reaction)	Ascorbic acid esters, Butyl hydroxyl toluene (BHT), Butyl hydroxyl anisole (BHA), Nor dihydrogualeic acid (NDGA), Tocopherols.
3. Synergists - (Increase effectiveness of other antioxidants.	Ascorbic acid, citric acid, Tartaric acid, phosphoric acid.
4. Chelating agents - forms complexes & inactivate metal those are responsible for oxidative degradation of drug	
Ex :-	EDTA, Disodium edentate, Disodium ca. edentate

* BUFFERING AGENT - USED IN CONCⁿ OF 0.1 to 5 %

EX :- Acetic acid, Adipic acid, Benzoic acid, Citric acid.

* Tonicity adjusters - NaCl, Mannitol.

* Stabilizers - Glycine, Niacinamide.

* Emulsifying agent - Lecithin

* Suspending agents - Gelatin, Pectin, Methyl cellulose, PEG - 4000

HEPA (High Efficiency Particulate Air) filter -
→ Removes out particles of 0.3 microns.
→ Air velocity - 100 + 20 ft/min.

Efficiency is checked by ★ Bubble point ★ DOP test

Freeze - Drying (Lyophilization)

→ When drug is unstable in solⁿ form or thermolabile.
→ Generally F.P - 50 °C (should be done below the product collapse temperature)

PARENTERALS

CONTAINERS USED FOR PARENTERALS ARE :-

★ GLASS CONTAINER			
Type	Description	Test	General Use
I	Highly resistant, borosilicate glass	Powdered glass	Buffered & unbuffered aq. sol ⁿ , sterile product
II	Treated soda lime glass	Water attack	Buffered aq. sol ⁿ below pH 7, dry powder & oleaginous sol ⁿ .
III	Soda lime glass	Powdered glass	Dry powder and oleaginous solution.
IV	General purpose soda lime glass	powdered glass	Not for parenterals.

★ PLASTIC CONTAINER	
Type	Uses.
Poly-propylene	Widely used due to high M.P
Poly-ethylene	Ophthalmic products
Polystyrene	Plastic syringes
Flexible PVC	Bags for I.V solution
Nylon & silicon rubber	I.V catheters

★ RUBBER CLOSURE	
Type	Example
Elastomers	Natural rubber, Butyl rubber, Neoprene.
Vulcanizing Accelerators	Sulfur.
Activator	2-mercaptobenzothiazole
Fillers	ZnO, stearic acid
Antioxidant	Carbon black, Clay Lime stone. Dilauryl thiodipropionate

★ DEVICES USED
 → I.V catheters - Nylon & silicon
 → Hypodermic needle - stainless steel

★ PYROGENS
 → these are lipopolysaccharide
 → these testing is not imp. in Ophthalmic.

★ ISOTONIC SOLUTION
 when osmotic pressure of solⁿ is equal to 0.9% NaCl or 1.9% Boric acid.

EVALUATION OF PARENTERAL PRODUCTS

LEAKER TEST (packaging integrity test)	- Only for ampoules, 1% Methylene blue dye & vacuum used.
CLARITY TEST (particulate matter test)	- Instrument like - light scattering (Nephelometer) - light absorption. - electrical resistance (counter counter).
PYROGEN TEST	- Fever response in rabbits Limulus test (Lal test) - 5 to 10 times more sensitive than rabbit test. Based on gelling of pyrogenic prep ⁿ .
STERILITY TEST	- Method → Membrane filtration. → Direct inoculation. Incubation - for 2 weeks at 20 to 30 °C

PARENTERALS

NOTE :-

- ★ Positive pressure is normally found in Aseptic & sterile area (class 100 area)
- ★ Aseptic process validation - sterile media fill
"broth fills"
- ★ Solid content of parenteral suspension usually range betⁿ. 0.5 to 5% (Dextrose) may go to high as 30%



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TABLET

✓ Important points to be remembered

- (a) In chewable tablet mannitol used as diluents. No disintegrating agent required.
- (b) Nitroglycerin & Isosorbide gives sublingually.
- (c) Diluents also known as fillers.
- (d) Lactose is mostly used diluents.
- (e) 5-80% diluents can be used.

✓ Important diluents trade name

Diluents	Trade name
Dextrase :-	Cerebase
Sucrose :-	Sugar-tab, Dipac, Nutab.
Microcrystalline Cellulose	Avicel
Starch :-	Emdex, Celutab, Sta Rx 1500
Modified Starch :-	Brimogel, Explotab.

Excellent Lubricant	Excellent Antihednant	Excellent Glidant
<ul style="list-style-type: none"> ▷ Magnesium Stearate ▷ Calcium Stearate 	Talc, Corn starch	Colloidal Silica [• Cab-o-sil • Aerosil • Syloid]

✓ Question asked from parts of Tablet machine :-

- ✓ Hopper → For holding & feeding.
- ✓ Dies → Define size & shape
- ✓ Punches → Compression of granules with die.
- ✓ Cambrack → Guide Movement of Punches.
- ✓ Turnets → Hold upper & lower punches.
- ✓ Feeding machine → Used for mixing.
- ✓ Die table → Position holding the dies.

Note :-

- ▷ Shape factor should be 6 for granule.
- ▷ Surface area is measured by
 - ✓ Gas adsorption method.
 - ✓ Air permeability method.

• Strength is determined by plating of granules between anvils (Tumbler test).

• Slugging process is related to dry granulation.

✓ Note :-

- # Lactose gives mullard reaction with $-NH_2$ containing drug.
↓
Due to formation of furfuraldehyde
- # Mannitol has cooling effect due to negative heat of solⁿ.
- # Sorbitol is optical isomer of mannitol.
- # Aspirin is directly compressed tablet so no need of diluents.
- # Saccharin 500 times sweeter than sugar but carcinogenic so currently Aspartame is used it is 200 times sweeter than sugar.
- # Flavouring agents used 0.5 to 0.75% w/w

Flow Properties	Angle of repose
Excellent	25-30
Good	31-35
Fair	36-40
Passable	41-45
Poor	46-55
Very poor	56-65
Very very poor	> 66

Carr's Index
Compressibility Index
15% → Excellent
18% → Good
25% → Poor
100% → Very poor

Comparison Box ↓

Variables	Disintegration	Dissolution
✓ Temperature	37 ± 2°C	37 ± 0.5°C
✓ Speed	28-32 cps	50-100
✓ Distance below surface of liquid	2.5 cm (25mm)	2.3-2.7 cm (23-27mm)

Dissolution Apparatus I (Basket) & II (Paddle) → U.S.P & B.P

Reverse in IP

TABLET

* Mostly asked questions from Tablet defects / Coating defects

Types	Defects
① Mottling	Unequal distribution of colour
② Orange peel effect	Inadequate spreading of coating material
③ Capping	Separation of top or bottom due to air entrapment
④ Lamination	Separation of tablet into two or more layers
⑤ Sticking	Adhesion of granulation material to the die wall
⑥ Picking	Adhesion to the face of punch
⑦ Cracking	when internal stress exceeds tensile strength of film.

* Important points in Tablet coating

▷ Coating is done by four types Sugar, Film, Enteric & specialised coating.
▷ Alcoholic soln is used for sealing [Alcoholic soln of zinc & shellac].
▷ Polishing is done by Bees wax, Paraffin & Carnuba wax mostly.
▷ Fluidized bed process also known as Air suspension coating.

* Mostly asked examples in questions:-

▷ Diluents :-	Lactose, Dextrose, Mannitol, Sorbitol, Avicel, Dicalcium phosphate.
▷ Granulating Agent	water, starch, mucilage, tragacanth mucilage, alcohol acetone
▷ Binders & adhesive	Starch, sucrose, Gelatin, Cellulose derivative, Cross linked PVP.
▷ Lubricants :-	Magnesium & calcium stearate
▷ Antiadherants	Talc, Corn starch etc.
▷ Glidants :-	Colloidal silica, Corn starch
▷ Disintegrants	Alginate, PVP, Primogel, SLS, Vegum (Magnesium aluminium silicate)
▷ Superdisintegrants	Sodium Starch glycolate, Cross povidone, Croscarmellose.
▷ Sweetening Agents	Mannitol, Aspartame, Saccharine.
▷ Subcoating binders	Gelatin, Sugarcane, PEG
▷ Subcoating Dusting Powders	TiO ₂ , Talc, CaCO ₃
▷ Polishing Agent :-	Bees wax, Paraffin, Carnuba wax.
▷ Film Coating :- materials	HPMC, MHEC, PVP, PEG, Acrylated Polymer (Eudragit E)
▷ Enteric Coating :- materials	phthalate derivatives, Eudragit (L.S)
▷ Plasticizers	Tween, Span, Castor oil, PEG, Glycerin
▷ Colorants	Iron oxide pigment (Red colour), Titanium Dioxide (white), Lakes, Carmel, Carotenoids etc.

TABLET

Evaluation of Tablets:-

Evaluation Test	Test Parameters	Determined By	Standard values & other points to be remembered											
① General Appearance	✓ Size & Shape ▷ Compressed tablets shape & dimensions	By tooling during the compression process	± 5 %											
	▷ Crown thickness of tablet	By micrometer												
	▷ Total crown thickness	By vernier callipers												
	✓ Colour	By • Reflectance spectrophotometry • Tristimulus - Colorimetry • Micro-reflectance photometry	Colour uniformity testing											
② Tablet-Hardness	✓ Hardness / Strength / diametric compression test	By ✓ Monsanto tester ✓ Strong Cobb tester ✓ Pfizer & Erweka tester	minimum 4kg											
③ Friability	✓ Drug loss during transportation. Note Tablet falls - 6 inch Speed - 25 rpm Time - 4 min. Total revolution - 100	By Roche friabilator	Not more than 1% (I-P) $\% \text{ friability} = \frac{\text{Initial wt} - \text{Final wt}}{\text{Initial wt}} \times 100$ # 0.5-1% (USP)											
④ Weight Variation	Randomly 20 select →		Passed batch											
	if 18 tablets remain 90mg to 110mg	if 2 tablets remain within 80 to 120mg												
	# Average wt. of Tablets													
	<table border="0"> <tr> <td>IP</td> <td>USP</td> <td></td> </tr> <tr> <td>80 mg / Less</td> <td>130 mg or less</td> <td>± 10 %</td> </tr> <tr> <td>80 - 250 mg</td> <td>130 - 324 mg</td> <td>± 7.5 %</td> </tr> <tr> <td>More than 250 mg</td> <td>More than 324</td> <td>± 5 %</td> </tr> </table>	IP	USP		80 mg / Less	130 mg or less	± 10 %	80 - 250 mg	130 - 324 mg	± 7.5 %	More than 250 mg	More than 324	± 5 %	
IP	USP													
80 mg / Less	130 mg or less	± 10 %												
80 - 250 mg	130 - 324 mg	± 7.5 %												
More than 250 mg	More than 324	± 5 %												
⑤ Uniformity of content	30 taken → 10 analysed if 9 → 85-115% if 10 → 75-125% Other 20 is not 85-115% range outside	By Analytical Technique	Passed Batch											

SOLUTION

- Aromatic water → volatile material in H_2O
- Syrup → contain 85% w/v sucrose
or 66.7% w/w sucrose
- Elixir → sweetened hydroalcoholic solⁿ
- Spirits → alcoholic or hydroalcoholic solⁿ
- Linctuses → relief of cough
- Liniment → alcoholic & oil liquid, rubbing on to the affected area
- Lotions → aqueous, alcoholic & oily liquid, without rubbing on the affected area

Solubility as per IP

- Very soluble → less than 1
- Freely soluble → 1-10
- Soluble → 10-30
- Sparingly soluble → 30-100
- Slightly soluble → 100-1000
- very slightly soluble → 1000-10,000
- Insoluble → More than 10,000
- Solubility study can be determined at 30°C
- Henry's law → Measure the solubility of gases in solvents

* Sweetening agent ⇒ 1) Saccharine (Approx 250-500 times as sweet as sugar)

2) Aspartame (Approx 200 times than sucrose)

⇒ Aspartame is methyl ester of Aspartic acid + phenylalanine

* 5-10% ethanol used in formulatⁿ for antimicrobial agent

* Monosodium glutamate has been widely used in food industry to mask the metallic taste.

* Taste property improves by sodium bicarbonate, gluconate, lactose

SUPPOSITORIES

- Solid dosage form
- Insertion into body cavity rectum, vagina, urethra

- Types of suppositories:
- Rectal suppositories
 - Vaginal suppositories
 - Urethral suppositories (Bogies)
 - Nasal suppositories
 - Ear cones

- Ideal suppository base:
- Melt at body temperature
 - Inert, non irritating & non sensitizing
 - If the base is fatty than
 - * Acid Value - Below 0.2
 - * Saponification Value - 200 to 245
 - * Iodine Value - Less than 7

Suppository base:-

Suppository base	Example
Oleaginous (fatty) base	Cocoa butter (Theobroma oil), Synthetic fats Hydrogenated palm kernel oil
Aqueous Base	Glycerinated gelatin (14% gelatin + 70% glycerine + H ₂ O)
Emulsifying Base	Maseq esterinum, Ultepsal, Massupol

- ★ Cocoa butter
- Most widely used suppository base
 - Melting point = 30-35°C
 - Iodine value = 34-38
 - Acid value not higher than 4
 - Polymorphism - $\alpha, \beta, \beta', \gamma$
 - Melting point of α = 24°C
 - Melting point of β' = 28-31°C
 - Melting point of β = 34-35°C
 - Melting point of γ = 18°C
 - β form is more stable

- Factors affecting drug absorption from rectal suppositories:-
- Physiological factor - pH of rectal mucosa.
 - Physicochemical characteristic of drug -
 - Particle size matter - \downarrow se absorption
 - Presence of surfactant - \uparrow se (NaI)
 - Physicochemical characteristic of the base & adjuvant
 - Melting point of base
 - cross linking $\propto \frac{1}{\text{Release of drug}}$
 - Viscosity $\propto \frac{1}{\text{Release of drug}}$

SUPPOSITORIES

Stability problems in suppositories :-

▷ Water

- present accelerate oxidation of fats.
- Evaporate cause crystallization

▷ Incompatibility

- silver salt, Tannic acid, Aminopyrine, Sulphonamide incompatible with PEG base
- penicillin → stable in cocoa butter
→ unstable in PEG

▷ Viscosity :-

- If less - Non-uniform storage problem
- For 1% viscosity 2% monoesterate

▷ Blooming

- White powder deposited at surface

▷ Brittleness

▷ Rancidity & Antioxidant

- Calculated by peroxide value
- Example of antioxidant
 - α , β tocopherol (VIT E)
 - Gossypol
 - sesamol
 - propyl gallate

- ⇒ Polyoxyethylene saccharin fatty acid esters (Tween) & polyoxyethylene stearates (Myrij) are example of water dispersible suppository base
- ⇒ The tensile of suppository is measured by - Breaking strength.
- ⇒ PEG base is used in the preparation of vaginal suppositories.
- ⇒ Cetyl alcohol is used as coating material in the preparation of coated suppositories.
- ⇒ Silver nitrate - increase the melting point of cocoa butter.
- ⇒ Triglycerides of saturated vegetable oil is the chemical nature of witepsol used as a suppository base.

AEROSOLS

* TYPES OF AEROSOLS

- 1) Two phase (gas & liquid)
- 2) Three phase (gas, liquid & solid or liquid)

* Classification of Aerosols

- 1) Metered dose inhaler
- 2) Powdered drug
- 3) Nebulizer
- 4) Dental spray
- 5) Skin spray

* Classification of Propellant

For oral & Inhalational	Topical
Fluorinated Hydrocarbons → • Trichloromonofluoromethane (011) • Dichlorodifluoromethane (012) • Dichlorotetrafluoroethane (0114)	Hydrocarbons → propane, Butane, Isobutane compressed gases → Nitrous oxide, Nitrogen, CO ₂

* HYDROCHLOROFLUOROCARBONS & HYDROFLUOROCARBONS (# low ozone depleting)

- chlorodifluoromethane (CHClF₂, P₂₂)
- Trifluoromonofluoroethane (CH₃CH₂F, P_{134a})
- chlorodifluoroethane (CH₃CClF₂, P_{142b})
- Difluoroethane (CH₃CHF₂, P_{152a})
- Heptafluoropropane (CF₃CHFCF₃, P₂₂₇)

* NAMING OF CFC

- 1st digit represents - [nC-1]
- 2nd digit represents - [nH+1]
- 3rd digit represents - [nF]

* CALCULATION

Dalton's law :-

$$P = P_1 + P_2 + P_3$$

Raoult's law :-

$$P = (x \times P_0)$$

AEROSOLS

* COMPONENTS OF VALVE

- Ferrule or mouth cap \Rightarrow Al, Brass
- Valve body or housing \Rightarrow Nylon, Al
(Opening 0.0132 - 0.08 inch)
- Stem \Rightarrow Nylon, Al
- Gasket \Rightarrow Buna-N & Neoprene
- Spring \Rightarrow Stainless steel
- Dip tube \Rightarrow Polypropylene & polyethylene

* EVALUATION OF AEROSOL

1. Flame Projection
 - extension of an open flame
 - product is spray 4 sec. into a flame
 - Measured by ruler
 - Determine flammability & combustibility
2. Flash point
 - Tag open cup apparatus
 - Temp - 25°C
3. Vapour pressure \rightarrow pressure gauge
4. Density \rightarrow Hydrometer & pycnometer
5. Moisture content \rightarrow Karl-Fisher method & Gas chromatography
6. Identification of propellants \rightarrow Gas chromatography
IR Spectroscopy
7. Foam stability
 - visual evaluation
 - Rotational viscometer
8. Particle size
 - cascade impactor (0.1 - 30 μ m)
 - light scatter decay

SUSPENSION

- Thermodynamic stable, heterogeneous system
- Size of the particle $\rightarrow 0.5 - 5 \mu\text{m}$
- Prolonged action \rightarrow Protamine zinc - Insulin suspension
 \rightarrow Procaine penicillin G suspension
- Dilute suspension $\rightarrow 2-10 \text{ w/v solid}$
- Concentrated suspension $\rightarrow 50\% \text{ w/v solid}$

Flocculated Suspension	Deflocculated Suspension
Particles form loose aggregate and form network like structure. Rate of sedimentation is high sediment is easy to redispense sediment is loosely packed & not form a hard cake	Particle exists as separate entities. Rate of sedimentation is low. Sediment is difficult to redispense sediment is very closely packed & form a hard cake

- * For an ideal suspension, the sedimentation volume should be \rightarrow equal to one
- * Preparation of structured vehicle \rightarrow Methyl cellulose is used
- * colloidal material is used common in the prepⁿ of structured vehicle \rightarrow hydrophilic
- * structured vehicle is included in the formulation suspension, in order to \rightarrow prevent the sedimentation of particles
- * In the extemporaneous prepⁿ of suspension levigation is used to \rightarrow Reduced particle size
- * DLVO Theory is \rightarrow Used to explain stability of emulsions
Used to explain Brownian movements of colloids
- * The amount of vehicle needed to produce pourability is denoted by \rightarrow Flow point

SUSPENSION

* Formulation of Suspension

<p>Flocculating agents</p> <p>Eg:- SLS, Tween, span, carbonyl wax</p>	<p>Suspending agent, Thickening agent</p> <p>Eg:-</p> <ul style="list-style-type: none"> Inorganic agent → clay, Al(OH)₃ Synthetic → carbomer, colloidal SiO₂ polysaccharide → Acacia, starch, Talc, MC, HPMC 	<p>Protective colloids</p> <p>Eg:-</p> <p>Gelatin, Natural gum, cellulose derivatives</p>	<p>Wetting agent</p> <p>Eg:-</p> <ul style="list-style-type: none"> Alcohol in tragacanth mucilage Gelatin in sodium alginate 	<p>Preservative</p> <p>Eg:-</p> <p>Benzoic acid, Methyl paraben, Sodium benzoate</p>	<p>Organoleptic additives</p> <ul style="list-style-type: none"> Sweetening agent colouring agent Flavouring agent
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* Zeta potential of suspension is helpful to find out the → **stability of suspension**

* Stoke's law ⇒ **Rate of Sedimentation is expressed**

$$\frac{dc}{dt} = \frac{D^2(d_1 - d_2)g}{18\eta}$$

* Sedimentation volume = 1 (No sedimentation)
sedimentation volume = 0 (complete sedimentation)

* Stokes law applicable to deflocculated system - where particle settle independently

* Deflocculated suspension shows → Newtonian flow

* flocculated suspension shows → Non-Newtonian flow

* The protamine - zinc suspension is prepared by a method viz. → Altered pH precipitation

* Which of the wetting agent used in parenteral suspensions → SLS

* Carboxy vinyl polymer is a → Carbomer

SUSPENSION

* Zeta potential →

- 1) The measurement of zeta potential gives →
 - Change on particles,
 - Stability of suspension
- 2) Calcium hydrogen phosphate is dissolved in H_2O . The apparent zeta potential initially is → Negative
- 3) The determination of change in the particle size of suspension is used → Zeta meter

* Dilute → 0.5 to 2% solid - Stokes law apply

* Concentration → 5 to 10% solid - Hindrance occurs

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CAPSULE

# Not Used for Administering - <ul style="list-style-type: none"> • extremely soluble material <u>cause</u> → Irritation (KCl, KBr, NH₄Cl, I) • Highly efflorescent <u>cause</u> → softness of shell • Highly Deliquescent <u>cause</u> → bitterness of shell 	# Used to - <ul style="list-style-type: none"> → Mask the taste → Mask the odour.
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GELATIN :- Heterogeneous product, derived by Hydrolytic extraction of animal collagen.

TYPES OF GELATIN

TYPE	SOURCE	PROCESSING	ISO-ELECTRIC POINT
Type - A	Pork skin	Acid Processed	PH - 9
Type - B	Bones	Alkali Processed	PH - 4.7

* HARD GELATIN CAPSULE (HGC) :-

Components of HGC shell

- Capsule shell - Gelatin
- Dyes - Indigo, erythrosine.
- Opacities - TiO₂
- Plasticizer - sorbitol, Glycerine
- Preservative - Methyl paraben & Propyl paraben (4:1)

size & filling Capacity of gelatin			
Capsule size	Size (inmm)	Vol. (ml)	Wt. (mg)
000	26.3	1.37	950
00	23.7	0.95	650
0	21.8	0.75	450
1	19.2	0.55	300
2	18.3	0.40	250
3	15.3	0.30	200
4	14.7	0.25	150
5	11.9	0.15	100

Method of Manufacturing of empty HGC shell.

- Centrifugal casting method.
- Dip pin method (mostly used) (Thickness depend on speed & time)

STEPS	DESCRIPTION
⇒ Dipping	Temp. of pin = 22°C, Sol ⁿ temp = 50°C Time of Dipping = 12 sec.
⇒ Spinning	Pins are rotated.
⇒ Drying	By use of Dry air & Dehumidification.
⇒ Stripping	By Bronze jaws.
⇒ Trimming	By stationary knives.
⇒ Joining	Cap & Body are joined.
⇒ Polishing	By Polymer.

CAPSULE

Process of filled HGC :-

- * Preparing the formulation.
- * Selection of the size of capsule
- * Proper mixing & Blending.
- * Filling the capsule shells.
- * Capsule sealing & shooting.
- * Cleaning & polishing.

Method of Cleaning & Polishing

- * Pan Polishing - Accele cota tablet, coating pan used.
- * Cloth dusting.
- * Brushing
- * Erweka KEA Machine.
- * Scidenader PM 60 Machine.

Capsule filling Machine

- Hofliger & Kay → Bulk powder
- Perry → Bulk powder
- Farmatic → Bulk powder.
- Macofar → Bulk powder.
- ELILLY → Pellet, granules
- Zanasi → Powder, pellet, granules
- Osaka → Based on vibration prin.
- MGI₂ → Continuous filling of powder.
- RotoffU → Pellets.

Capsule Weighing Machine

- Vericap - 1200
- Rotoweigh.

Condition for Capsule

- Storage = 100°F (35°C)
- Area temp. = 22°C (Processing)
- Humidity = 30 - 45 %
- Moisture Content = 12 - 16
- Gelatin : Plasticizer = 1 : 0.4

SOFT GELATIN CAPSULE :-

Components are :-

- Gelatin - Bloom strength = 150-120, Viscosity = 25-45 millipoise
Iron content = NMT 15 PPM
- Plasticizers - sorbitol, glycerin, PEG. [Gelatin : Plasticizer
1 : 0.8]
- Preservatives - 0.2% Methyl paraben : Propyl paraben
4 : 1
- Opacifiers - TiO₂
- Fumaric acid - 1% ↑ acid solubility & ↓ aldehyde tanning.
- Sugar - 5%
- Essential oil - 2%
- Formalin - Retards dissolution of gelatin.
- Coating material - Salol, CAP, shellac.
- SO₂ - 0.15% of gelatin (Prevent Decomposition).

CAPSULE

Evaluation of Capsule :-

① Invariability Test

Weight Variation	
Avg. wt. of caps.	% deviation
less than 300 mg	10 %
300 mg or more	7.5 %

Content Uniformity
test pass if, all 30 capsule are within 75-125% range & NLT 27 of 30 are within 85-115% range.

② Dissolution test - Rotating Basket type.

③ Disintegration test:

- For HGC = 30 min.
- For SGC = 60 min.

④ Stability test :-

- ⇒ Shell integrity test.
- ⇒ Determination of self life.

Note :-

- ★ Soft spot (Blooming) - site at which capsule lie next to the tray or against another capsule.
- ★ Bloating - At high humidity (60%) capsule size increases.
- ★ Foreign capsule - Unfilled capsule.

STERILIZATION

process designated to produce a sterile state.

* Some basic terms used: -

- 1) Sterility: Condition of freedom from living organisms. (absolute)
- 2) D-value (Decimal Reduction Time): - Time or dose required for microbial population to decline by one decimal point.
- 3) Z-value (Thermal destruction value): - degree of temperature required for 1 log reduction in D value.
- 4) F-value: time in minutes at a specific temp. (usually 250°F / 121°C) needed to kill a population of cells or spores.
- 5) Holding Time: Time required for killing the organism
- 6) Thermal Death Time: (TDT): shortest time needed to kill all organisms in a microbial suspension.

Sterilization Methods:

(A) Sterilization by HEAT:

1. Dry heat Sterilization:

used for: oils, petroleum jellies, talcum powder, glass & metallic apparatus.

* Microorganisms more resistant to dry heat than moist heat.

(a) Flaming:

used for: forceps, needles, scalpels, metal spatulas etc.

Bunsen flame to the surface for 5 seconds.

(b) Hot air oven:

180 minutes at 180°C
or 45 minutes at 260°C.

MoA: Destruction by means of heat & it is an Oxidation Process.

For glasswares & oily solⁿ: 160°C for 1 hour.

2. Moist heat Sterilization:

MoA: Coagulation of proteins in the cell.

Used for: porous materials such as cotton wool, stoppers, paper, cloth wrappers, bundles of surgical linen etc.

Temp °C	Pressure	Holding time (in min.)
115-116	10	30
121-123	15	15
126-129	20	10
134-138	32	3

STERILIZATION

(3) Tyndallization :

- Also known as 'fractional sterilization'
- devised by JOHN Tyndall
- 80°C for 1 hour or at 100°C for less time on 3 successive days.
- Kills dormant, resistant spores also
- No special apparatus required.

(5) Pasteurization :

- devised by Pasteur.
- It only reduces the viable count by 97 to 99%, but does not kill spores.
- applied to milk in dairies.

(6) Dry heat above 150°C.

bacterial spores passing through an extremely heated furnace can be sterilized in 0.4-0.6 sec. at 225°C.

(4) Heating with a bactericide:

- Bactericide are more effective at high temp than low temp.
- official in B.P. & consists of heating at 90-100°C for 30 minutes.

For injections:

Chlorocresol	0.2% w/v
Phenylmercuric acetate	0.002% w/v
Benzalkonium chloride	0.01% w/v
Thiomersal	0.01% w/v

* Note:

More uses of UV radiation

- i) inactivate viruses & bacteria in vaccine production.
- ii) also used for irradiation of incoming and/or internal air of the sterile filling areas of antibiotic.
- iii) prevent cross infection of hospitals & schools etc.

[B.] Sterilization by Radiation :

→ Also known as 'Cold sterilization'.

MOA → interference in the metabolism of the cell.

(1). Ultraviolet light :

- 2537 Å possess greatest activity in destroying m.o.
- reduction of air borne contamination & maintenance of aseptic areas & rooms.

MOA: UV light absorbed by nucleic acids of the cell wall where it does the greatest damage.

Uses: to destroy the viruses of homologous serum jaundice in blood.

(2). Ionizing Radiation:

Produced

↳ directly → from charged particles
↳ indirectly " gamma rays

→ Have highest penetrating powers.

MOA → destroy microbes by stopping reproduction as a result of lethal mutations.

- Uses: i) preservation of foods.
ii) injection products of antibiotics like benzylpenicillin, streptomycin sulphate

STERILIZATION

* Note:

UNIT for measurement of radiation doses:

- 1) Roentgen (r) → fundamental unit for radiation doses.
↳ It is defined in terms of the number of ionizations produced per unit vol. of air.
- 2) rad. → Another unit which represents an energy absorption of 100 ergs per gram of material.

[C]. Sterilization by Filtration:

- Used for sterilization of heat-labile solutions.
- * Official in G.P. for sterilization of certain injection solⁿ.

Efficient filters: should be able to retain *Serratia marcescens*.

Av. pore diameter = 0.75 μ m or less.

Commonly used filters are:

- porcelain filters, diatomaceous earth filters, asbestos pad filters, sintered glass filters, membrane filters & ultrafilters.

MoA: removal of microbes (physical) from sample

[D]. Chemical Processes of Sterilization (Gaseous Sterilization)

① Ethylene Oxide: $\text{H}_2\text{C}-\text{CH}_2$

- Simplest cyclic ether
- highly inflammable
- may be explosive at conc > 3% in air.
- made safe by mixing with inert gases CO_2 etc.

STERILIZATION

② Ozone:

→ Pleasant smelling but
irritant & toxic gas
MST
→ general protoplasmic oxidant

Uses:

- i) disinfection of water
- ii) preservation of foods from spoilage during bulk storage

③ Formaldehyde:

Used for: fumigation of rooms & hospital blankets.

- Irritant & pungent
- kills all bacteria, including spores, rapidly.
- Inactivated by proteins & organic matter.

④ β -Propiolactone (BPL):

- highly bactericidal
- conc used - 5 mg/litre or less.

Uses:

- i) decontaminating rooms, chamber & buildings.
- ii) highly virucidal but - carcinogenic to animals.



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